

The Reaction of 3-Oxo- Δ^4 -steroids with Isocyanuric Chloride

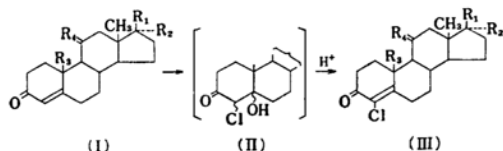
By Fumikazu MUKAWA

(Received May 13, 1959)

It has been reported that 3 β -substituted- Δ^5 -steroids gave the corresponding 5 α -chloro-6 β -hydroxy compounds on treatment with isocyanuric chloride, and that 3-oxo- Δ^5 -steroids and their ethyleneketal gave the 6 β -chloro-5 α -hydroxy compounds¹⁾.

Now, similar studies on the reaction of 3-oxo- Δ^4 -steroids (I) with isocyanuric chloride are reported herewith.

Treating the model compound progesterone with isocyanuric chloride in acetone-acetic acid (or perchloric acid) yielded a chloro compound, having an ultraviolet absorption typical for the 4-substituent-4-en-3-one system ($\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ , log ϵ 4.12)²⁻⁴⁾. The identity of the product with 4-chloroprogestosterone was established by its melting point and the infrared spectrum. Therefore, it appears that 4-chloroprogestosterone (Ic) is formed through an intermediate 4 ξ -chloro-5 ξ -hydroxy compound (II):



- (a) $R_1 = \text{OAc}$, $R_2 = \text{H}$, $R_3 = \text{CH}_3$, $R_4 = \text{H}_2$
 (b) $R_1 = \text{OCOC}_2\text{H}_5$, $R_2 = \text{H}$, $R_3 = \text{CH}_3$, $R_4 = \text{H}_2$
 (c) $R_1 = \text{COCH}_3$, $R_2 = \text{H}$, $R_3 = \text{CH}_3$, $R_4 = \text{H}_2$
 (d) $R_1 = \text{COCH}_3$, $R_2 = \text{OH}$, $R_3 = \text{CH}_3$, $R_4 = \text{H}_2$
 (e) $R_1 = \text{COCH}_3$, $R_2 = \text{OAc}$, $R_3 = \text{CH}_3$, $R_4 = \text{H}_2$
 (f) $R_1 = \text{OH}$, $R_2 = \text{CH}_3$, $R_3 = \text{H}$, $R_4 = \text{H}_2$
 (g) $R_1 = \text{COCH}_2\text{OAc}$, $R_2 = \text{OH}$, $R_3 = \text{CH}_3$, $R_4 = \text{O}$

Extension of this chlorination method to 3-oxo- Δ^4 -steroid hormones, led to 4-chloro-derivatives, several of which were characterized by their absorption maximum.

Since this work was completed, Oliveto et al. reported⁵⁾ the preparation of 4-chloro-

cortisone acetate by *N*-chlorosuccinimide and perchloric acid.

4-Chlorotestosterone acetate (IIIa) and 4-chlorotestosterone propionate (IIIb) exhibited greater anabolic and smaller androgenic effect in comparison to ad-ministeral testosterone propionate, and showed no inhibitory effects on the gonadotropic secretion upon intraspleenic ovarian transplantation⁶⁾.

4-Chloroprogestosterone (IIIc) and 4-chloro-17 α -acetoxyprogesterone (IIIe) showed no progestational activity.

Experimental*

5 α -Chloro-3 β -acetoxy-pregnan-6 β -ol. — To a mixture consisting of 1.0 g. of pregnenolone acetate, 0.5 g. of isocyanuric chloride and 30 ml. of acetone were added ten drops of 30% acetic acid. The whole was refluxed for 5 min. After cooling to room temperature, the mixture was poured with stirring into 10% aqueous sodium carbonate, and then extracted with ether. The extract was washed with water and dried over magnesium sulfate, after removal of the solvent, the residue was recrystallized from acetone-water to yield 0.6 g. of needles, m. p. 195°C (decomp.), $[\alpha]_D^{25} +36$ (c. 1.2).

Anal. Found: C, 67.26; H, 8.65; Cl, 8.42. Calcd. for $\text{C}_{23}\text{H}_{35}\text{O}_4\text{Cl}$: C, 67.37; H, 8.60; Cl, 8.65%.

Preparation of 4-Chloro-3-oxo- Δ^4 -steroids. — A solution of 0.5 g. of 3-oxo- Δ^4 -steroid and 0.3 g. of isocyanuric chloride in 30 ml. of acetone was refluxed for 1 hr. with 3 ml. of acetic acid or 10% perchloric acid. The reaction mixture was then poured into 10% sodium carbonate solution and extracted with ether. Washing of the extract with water, drying over magnesium sulfate, and evaporation of the solvent led to a residue which was purified by recrystallization or chromatography on florisil.

The following 4-chloro-3-oxo- Δ^4 -steroids were prepared by this method:

4-Chlorotestosterone acetate (IIIa), needles, m. p. 228°C (from acetone), $\lambda_{\text{max}}^{\text{EtOH}}$ 255 m μ (log ϵ 4.12);

4-Chlorotestosterone propionate (IIIb), needles, m. p. 164°C (from methanol), $\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ (log ϵ 4.12);

6) The parabiotic rat technique was reported: V. Cline, G. Baldratti and G. Sala, *Experientia*, 14, 94 (1958).

* All melting points are uncorrected. Optical rotation was measured in chloroform solution.

1) F. Mukawa, *J. Chem. Soc. Japan, Pure Chem. Sec. (Nippon Kagaku Zasshi)* 78, 452 (1957).

2) B. Camerino, B. Patelli and A. Vercellone, *J. Am. Chem. Soc.*, 78, 3540 (1956).

3) D. N. Kirk, D. K. Patel and V. Petrow, *J. Chem. Soc.*, 1956, 1184.

4) J. Ringold, B. Batres, O. Mancera and G. Rosenkranz, *J. Org. Chem.*, 21, 1432 (1956).

5) E. P. Oliveto, C. Gerold and E. B. Hershberg, *J. Am. Chem. Soc.*, 79, 3596 (1957).

4-Chloroprogesterone (IIIc), needles, m. p. 220°C (from acetone), $\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ (log ϵ 4.20);

4-Chloro-17 α -hydroxyprogesterone (III d), needles, m. p. 218°C (from methylene chloride-methanol), $\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ (log ϵ 4.11);

Anal. Found: C, 68.90; H, 8.10. Calcd. for $\text{C}_{21}\text{H}_{29}\text{O}_3\text{Cl}$: C, 69.29; H, 8.03%.

4-Chloro-17 α -acetoxyprogesterone (IIIe), needles, m. p. 194°C (from acetone after being separated by chromatography from benzene on florisil), $\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ (log ϵ 4.20);

Anal. Found: C, 67.85; H, 7.53. Calcd. for $\text{C}_{23}\text{H}_{31}\text{O}_4\text{Cl}$: C, 68.04; H, 7.70%.

4-Chloro-17 α -methyl-19-nortestosterone (III f), needles, m. p. 148°C (from ether-hexane after being separated by chromatography from benzene on florisil), $\lambda_{\text{max}}^{\text{EtOH}}$ 256 m μ (log ϵ 4.20);

Anal. Found: C, 70.93; H, 8.90. Calcd. for $\text{C}_{19}\text{H}_{27}\text{O}_2\text{Cl}$: C, 70.87; H, 8.45%.

4-Chlorocortisone acetate (IIIg), needles, m. p. 234°C (from acetone-*n*-hexane), $\lambda_{\text{max}}^{\text{EtOH}}$ 254 m μ (log ϵ 4.11).

Ultraviolet spectra and optical rotations were kindly determined by Mr. M. Sawai and Miss S. Imai.

The author wishes to express his sincere thanks to Dr. M. Chuman and Assistant Professor Y. Nomura of the University of Tokyo for their interest in this work. Thanks are also due to Dr. O. Kobayashi of Teikoku Hormone Manufacturing Co., Ltd., and Dr. K. Maekawa of Tokyo Jikeikai School of Medicine for bio-assay.

Tsurumi Research Laboratory of Chemistry
Shitanoya-machi, Tsurumi-ku, Yokohama